## Surface Energy Data for PCTFE: Poly(chlorotrifluoroethylene), CAS #9002-83-9

Source <sup>(a)</sup>	Mst. Type <sup>(b)</sup>	Data©	Comments <sup>(d)</sup>
Fox, 1952 <sup>(12)</sup> Lee, 1968 <sup>(131)</sup>	Critical ST Critical ST	$\begin{array}{l} \gamma_{\rm c}=31 \ mJ/m^2; \ 20^{\circ}C \\ \gamma_{\rm c}=31 \ mJ/m^2; \ no \ temp \ cited \end{array}$	Test liquids not known. Test liquids: water, glycerol, formamide, alcohols, and long- chain polyglycols.
Wu, 1971 <sup>(29)</sup>	Contact angle	$\theta_{W}^{Y} = 90^{\circ}; 20^{\circ}C$	1 305
Bee, 1993 <sup>(4)</sup>	Contact angle	$\theta_{W}^{"A} = 104^{\circ}, \ \theta_{W}^{"R} = 67^{\circ}, \ d\theta_{W} = 37^{\circ};$ no temp cited	
Bee, 1993 <sup>(214)</sup>	Contact angle	$\theta_{W}^{A} = 104^{\circ}, \ \theta_{W}^{R} = 77^{\circ}, \ d\theta_{W} = 27^{\circ};$ no temp cited	
Wu, 1971 <sup>(29)</sup>	Contact angle	$\gamma_{s} = 27.5 \text{ mJ/m}^{2} (\gamma_{s}^{d} = 23.9, \gamma_{s}^{p} = 3.6); 20^{\circ}\text{C}$	Test liquids: water and diiodomethane, by geometric mean equation.
Wu, 1971 <sup>(29)</sup>	Contact angle	$\gamma_{s}=30.1~mJ/m^{2}~(\gamma_{s}^{\rm d}=21.6,\gamma_{s}^{\rm p}=8.5);20^{o}C$	Test liquids: water and diiodomethane, by harmonic mean equation.
Wu, 1979 <sup>(<u>45</u>)</sup>	Contact angle	$\gamma_{\rm c} = 32.1 \text{ mJ/m}^2; 20^{\circ}\text{C}$	Test liquids not known, calculated by the equation of state method.
Kwok, 2000 <sup>(<u>166</u>)</sup>	Contact angle	$\gamma_{\rm c}$ = 28.9 mJ/m²; no temp cited	Re-calculated by equation of state method from data produced by Fox. $1952^{(12)}$ .
Schonhorn, 1966 <sup>(39)</sup>	From polymer melt	$\gamma_{s}=30.9~mJ/m^{2}~(\gamma_{s}^{\rm ~d}=22.2,~\gamma_{s}^{\rm ~p}=8.7);~20^{\rm o}C$	Measurement by capillary height of polymer melt extrapolated to $20^{\circ}$ C; polarity calculated from interfacial tension with PE by harmonic mean M = 1280.
Wu, 1971 <sup>(29)</sup>	From polymer melt	$\gamma = 31.1 \text{ mJ/m}^2$ ; 20°C	Direct measurement of polymer melt extrapolated to $20^{\circ}$ C.
Good, 1964 <sup>(16)</sup>	Calculated	$\gamma_{c} = 38.0 \text{ mJ/m}^{2}; 20^{\circ}\text{C}$	Estimated from molecular constants, using $u = 1.2$ debyes.
Lee, 1968 <sup>(131)</sup>	Calculated	$\gamma_{\rm s} = 26  {\rm mJ/m^2}$ ; no temp cited	Calculated from glass temperature of 318K.
Wu, 1968 <sup>(182)</sup>	Calculated	$\gamma_{\rm s} = 31 \text{ mJ/m}^2$ ; 20°C	Calculated from molecular constitution.

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