

### Surface Energy Data for PVA: Poly(vinyl acetate), CAS #9003-20-7

Source <sup>(a)</sup>	Mst. Type <sup>(b)</sup>	Data <sup>(c)</sup>	Comments <sup>(d)</sup>
Wu, 1968 <sup>(182)</sup>	Critical ST	$\gamma_c = 36 \text{ mJ/m}^2$ ; no temp cited	Test liquids not known.
Crocker, 1969 <sup>(111)</sup>	Critical ST	$\gamma_c = 37 \text{ mJ/m}^2$ ; no temp cited	Test liquids not known.
Wu, 1971 <sup>(41)</sup>	Critical ST	$\gamma_c = 33 \text{ mJ/m}^2$ ; 20°C	Test liquids not known.
McCafferty, 2000 <sup>(217)</sup>	Contact angle	$\theta_W^Y = 60.6^\circ$ ; no temp cited	Surface cleaned with light methanol wipe.
Roe, 1978 <sup>(223)</sup>	Contact angle	$\gamma_s = 42.9 \text{ mJ/m}^2$ ( $\gamma_s^d = 27.4$ , $\gamma_s^p = 15.4$ ); no temp cited	Test liquids not known.
McCafferty, 2000 <sup>(217)</sup>	Contact angle	$\gamma_s = 44.5 \text{ mJ/m}^2$ ( $\gamma_s^{LW} = 42.6$ , $\gamma_s^{AB} = 1.9$ , $\gamma_s^+ = 0.04$ , $\gamma_s^- = 22.3$ ); no temp cited	Test liquids: water, diiodomethane, formamide, ethylene glycol, and glycerin; acid-base analysis. Cleaned with methanol wipe.
Wu, 1969 <sup>(28)</sup>	From polymer melt	$\gamma_s = 36.5 \text{ mJ/m}^2$ ( $\gamma_s^d = 24.5$ , $\gamma_s^p = 12.0$ ); 20°C	Measurement by pendant drop of polymer melt extrapolated to 20°C; polarity calculated from interfacial tension with PE by harmonic mean. $M_w = 11,000$ .
Wu, 1970 <sup>(35)</sup>	From polymer melt	$\gamma_s = 36.3 \text{ mJ/m}^2$ ( $\gamma_s^d = 23.2$ , $\gamma_s^p = 13.1$ ); 20°C	Direct measurement of polymer melt extrapolated to 20°C.
Wu, 1971 <sup>(29)</sup>	From polymer melt	$\gamma_s = 36.5 \text{ mJ/m}^2$ ( $\gamma_s^d = 23.6$ , $\gamma_s^p = 12.9$ ); 20°C	Measurement by pendant drop of polymer melt extrapolated to 20°C; polarity calculated from interfacial tension with PE by geometric mean equation.
Lee, 1968 <sup>(131)</sup>	Calculated	$\gamma_s = 40 \text{ mJ/m}^2$ ; no temp cited	Calculated from glass temperature of 301K.
Wu, 1968 <sup>(182)</sup>	Calculated	$\gamma_s = 36 \text{ mJ/m}^2$ ; 20°C	Calculated from molecular constitution.
Wu, 1970 <sup>(35)</sup>	Calculated	$\gamma_s = 38.8 \text{ mJ/m}^2$ ; 20°C	Calculated from parachor and molecular weight.
Sewell, 1971 <sup>(193)</sup>	Calculated	$\gamma_s = 41.1 \text{ mJ/m}^2$ ; no temp cited	Calculated from parachor and cohesive energy.
Wu, 1982 <sup>(18)</sup>	Calculated	$\gamma_s = 39.1 \text{ mJ/m}^2$ ; 20°C	Calculated from cohesive energy density and solubility parameters.
Van Ness, 1992 <sup>(186)</sup>	Calculated	$\gamma_s = 37.0 \text{ mJ/m}^2$ ; 20°C	Calculated molten surface tension value, extrapolated to 20°C.
Pritykin, 1986 <sup>(199)</sup>	Calculated	$\gamma_s = 35.7 \text{ mJ/m}^2$ ; no temp cited	Calculated from cohesion parameters and monomer refractometric characteristics, equation 1.
Pritykin, 1986 <sup>(199)</sup>	Calculated	$\gamma_s = 38.0 \text{ mJ/m}^2$ ; no temp cited	Calculated from cohesion parameters and monomer refractometric characteristics, equation 2.
Surface-tension.de, 2007 <sup>(110)</sup>	Unknown	$\gamma_s = 36.5 \text{ mJ/m}^2$ ( $\gamma_s^d = 25.1$ , $\gamma_s^p = 11.4$ ); 20°C	No details available.